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A1g B2g A2u B1u Eg Eu none dxz py s.66 Chapter 13 Organometallic Chemistry a. reduces to the irreducible representations listed.200 13. dz2 pz dyz The representation . . so both the s and dz2 orbitals of nickel can interact with the zero-node group orbital on the left.

Chapter 13 Inorg | Coordination Complex | Inorganic Chemistry

13 Organometallic Chemistry of the Actinides By the end of this chapter you should be able to: recall that these compounds are very air- and water-sensitive; recall that the bonding has a significant polar character, especially in the +3 state, but

13 Organometallic Chemistry of the Actinides

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Chapter 13 - Organometallic Chemistry - Organometallic

Until fifty years ago, organometallic chemistry was restricted to a few oddities that were hard to understand. For example, $[\text{Ni}(\text{CO})_4]$ has been known since at least 1889, when it was used in the Mond process for production of ultrapure Ni. Ni is the only metal that will react directly with CO to produce a carbonyl, which is

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